

Date: _____

Course: _____

This sheet belongs to: _____

DICTOGLOSS - HOW STRONG IS AN ACID OR AN ALKALI?



1. Your teacher has just read a text about the pH, while you were taking notes about it. Glue the cards you have been provided in the same order as the information read by the teacher. Write next to each picture an explanation according to your notes. The boxes below the lines will help you.

a) _____

abbreviation	relative	solution
acidity	alkalinity	

b) _____

scale	change	times
acidic	alkaline	

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d)

solution	ion	lower
number	higher	

c)

hydrogen	ion	dissolve
break up	water	

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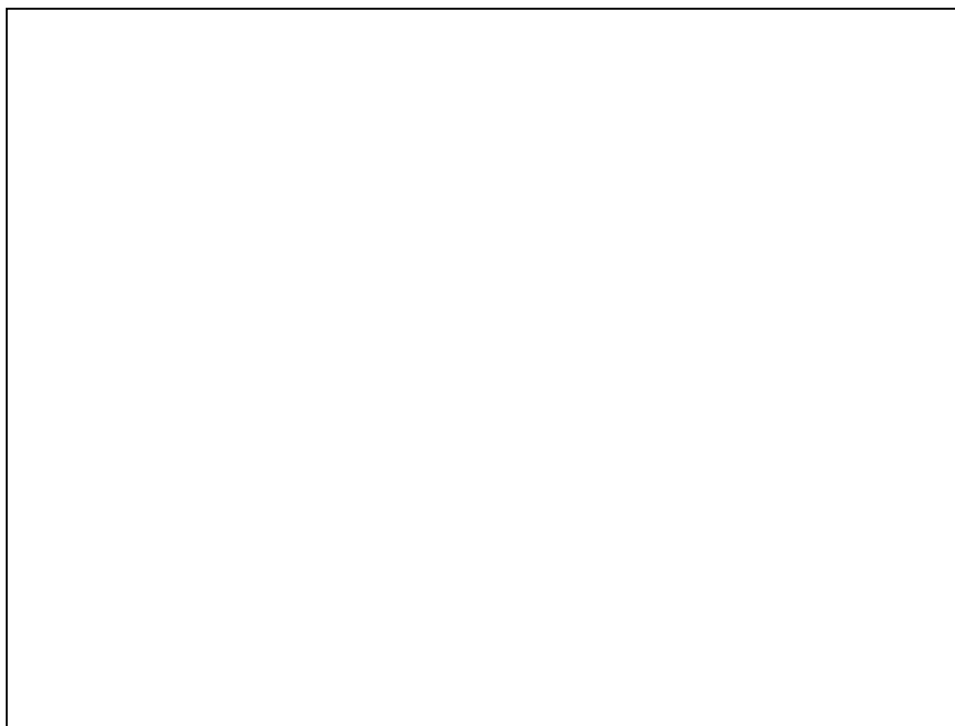
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e)

molecule	all	some	break up
		solution	

f)

find	strong	acid	alkali	weak
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g)

measure	ways	instrument
such as	impregnate	



2. Match the words of each of the four columns.

ammonia	H ₂ O	strong alkali	pH = 1
water	NaOH	weak acid	pH = 11
acetic acid	NH ₃	strong acid	pH = 7
sodium hydroxide	HCl	weak alkali	pH = 4
hydrochloric acid	CH ₃ COOH	neutral	pH = 14